Curriculum Map

Subject: Chemistry

Year: 11



	Autumn 1	Autumn 2	Spring 1	Spring 2	Summer 1	Summer 2
Content	C12 - Chemical Analysis (Recap)	C14 - The Earth's resources	C15 - Using our Resources	Revision of All Content	Revision of All Content	
Knowledge	Content: - Pure Substances and Mixtures - Chromatography - Gas Tests - Testing for lons - Instrumental Analysis Knowledge: - How chromatography can identify between pure and impure substances - How chromatography works - Tests for common gases - Common tests for positive ions - Tests for negative ions C13 - The Earth's Atmosphere Content: - History of the atmosphere - Changes to the atmosphere - Greenhouse Gases - Climate Change - Atmospheric Pollutants	Content: - Finite and Renewable energy sources - Safe Water - Treating water - Extracting metals - Life Cycle Assessments - Reduce, Reuse and Recycle Knowledge: - Know the difference between potable and pure water - How to purify water - How to make water safe for the environment - How to obtain potable water - Interpreting life cycle assessments - How reducing, reusing and recycling of materials decrease their environmental impacts Required Practical: - Analysis and purification of water samples	Content: - Rusting - Alloys - Polymers - Glass, Ceramics and Composites - Making Ammonia - Making fertilisers Knowledge: - The conditions required for rusting - Why metals are alloyed - How properties of polymers are determined by their monomers - Differences between thermosetting and thermosoftening - Comparing properties of glass, ceramics and composites - Why ammonia is an important compound - How to prepare fertiliser	- To focus on recapping key knowledge and re-address common misconceptions - Embed additional exam practice for each chapter - Focus on key aspects of required practicals	- To focus on recapping key knowledge and re-address common misconceptions - Embed additional exam practice for each chapter - Focus on key aspects of required practicals	

	Knowledge: - Understand how the atmosphere formed - Changes to the atmosphere over time - How greenhouse effect works - The importance of peer review - How to reduce carbon emissions - Problems caused by pollutants Required Practical - Identifying pure and impure substances by chromatography					
Skills	Distinguishing pure substances from impure by melting point Identifying formations Interpreting Chromatograms Evaluate modern instrumental methods with tradition analysis methods Interpreting results for flame emission spectroscopy Interpret evidence and evaluate different theories about Earth's early atmosphere Evaluate the quality of evidence in a report about global climate change	Distinguishing between finite and renewable sources Extracting and interpreting information from charts and graphs Determining the purity of water Evaluating the alternative biological methods of metal extraction Evaluating ways of reducing and uses of limited metal ores	Interpreting and evaluating the compositions and uses of alloys Evaluating industrial preparation of fertilisers against lab preparation of same compounds	Recalling important information Exam Technique Spacing Interleaving Elaboration Time management	Recalling important information Exam Technique Spacing Interleaving Elaboration Time management	
Key Questions	How can we use chemical tests to identify unknown substances?	How is human activity affecting the Earth's atmosphere?	How are we seeking to make sustainable use of the Earth's limited resources?	How do I revise for Chemistry?	How do I revise for Chemistry?	

Assessmen t	What are the advantages and disadvantages of using industrial methods of analysis? C13 Diagnostic Tests C13 End of Chapter Tests C13 ReAct Tasks	C14 Diagnostic Test C14 End of Chapter test C14 React Tasks Winter Mocks	C15 Diagnostic Test C15 End of Chapter Test C15 ReAct Tasks	Spring Mocks	GCSE exams	
Literacy/nu meracy/SM SC/Charact er	Numeracy: - Measuring the height of a chromatogram - Calculating the Rf Value of a chromatogram - Reading pie charts - Analysing graphs Keywords: Formulations, mixtures, chromatogram, retention factor, precipitates, atmosphere, carbon footprint, carbon storage, particulates, global dimming SMSC: - Working together to investigate chromatography - Evaluating evidence that demonstrates the rise of carbon dioxide leading to climate change	Numeracy: - Extracting and interpreting data from charts, graphs and tables - Using order of magnitudes Keywords: Bioleaching, life cycle assessment, blast furnace, non-renewable SMSC: - Working together to investigate the purity of water - Are there more sustainable ways of extracting metals?	Numeracy: - Balancing formulae Keywords: rusting, galvanising, sacrificial protection, alloys, steel, polymers, thermosetting, thermosoftening, ceramics, composites SMSC: - Are there more efficient ways of producing ammonia?			
Enrichment opportuniti es and futures	STEM Club Summer Fayre at Royal Institution of Chemistry Visiting the Science Museum The Wellcome Collection The Faraday Museum					